

SIMPLE SYNTHESSES OF THE ATLANTONES, OCIMENONES,
TAGETONES AND FILIFOLONE FROM ISOPRENE

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The formation of terpenes by step-wise addition of isoprene units has long been an objective of organic synthesis.¹ We describe here such synthesis of a range of terpenes by acylation of olefins and by Diels-Alder reactions, both catalysed by Lewis acids.

Addition of isoprene (0.13 mol.) in methylene chloride to senecieryl chloride (0.1 mol.) in methylene chloride containing stannic chloride (0.09 mol.) at -78° gave the crude chloro-ketone (8),[†] which was treated with lithium fluoride and lithium carbonate in dimethylformamide at 120° to give ocimenone² in 90% isolated yield, as a mixture of *cis* and *trans*-isomers³ (1:9) (5) and (6)*. *cis*- and *trans*-Atlantone⁵ (1) and (2), identical, but for optical rotation, with samples isolated from the essential oil of *Cedrus atlantica* were obtained from ocimenone (0.004 mol.) by addition of isoprene (0.016 mol.) in methylene chloride containing aluminium chloride (0.003 mol.) at 20° , in an isolated yield of 85%.

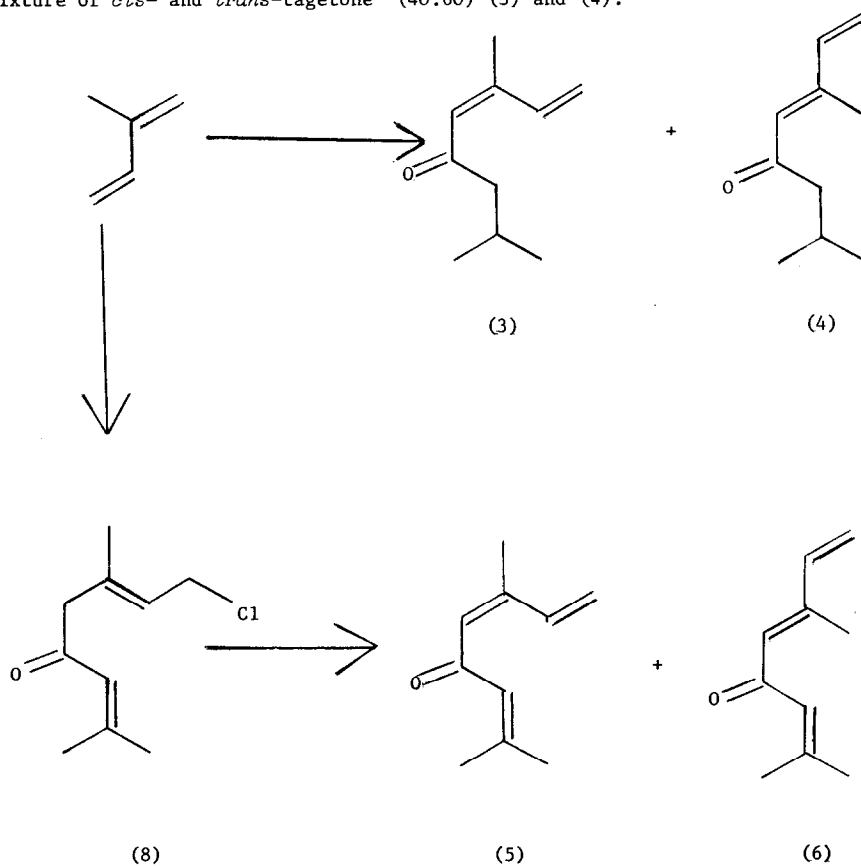
Treatment of ocimenone (0.004 mol.) in methylene chloride with aluminium chloride (0.003 mol.) at 40° resulted in the formation of filifolone⁶ (7) (80% conversion after 3h.), probably through the intermediate (11) suggested in the rearrangement of chrysanthene⁷ to filifolone.

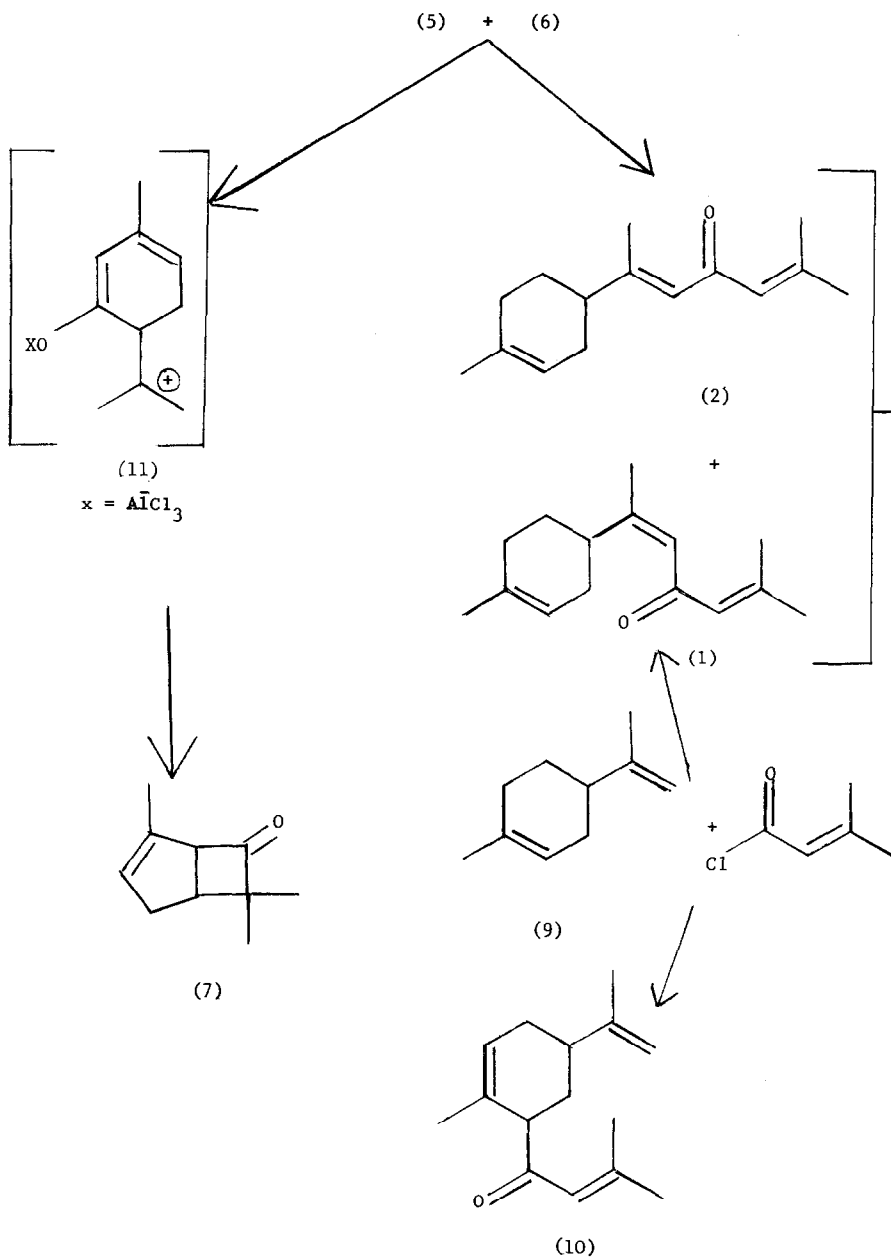
The *cis* and *trans*-atlantones (1:9) (1) and (2) were also synthesised by direct acylation and subsequent dehydrochlorination of limonene (9) (available by dimerisation of isoprene):

seneciyl chloride (0.05 mol.) in methylene chloride containing stannic chloride (0.045 mol.) cooled to -78° was added to limonene (0.05 mol.) in a mixture of methylene chloride and Freon 12 at -120° . The crude chloro-ketones were then treated with lithium fluoride and lithium carbonate in dimethylformamide at 120° to afford a mixture of isomers of which *cis* and *trans*-atlantone (1) and (2) were *ca.* 60% (glc) of the reaction mixture. The other major product (20%, glc) was the nonconjugated isomer (10).

† Although the chloro-ketones could be isolated and identified (m.s., n.m.r., i.r.), in a typical run the dehydrochlorination was carried out using the crude material.

* A similar sequence but using isovaleryl chloride as the acid chloride, furnished a mixture of *cis*- and *trans*-tagetone⁴ (40:60) (3) and (4).





References

1. A.F. Thomas, "The Total Synthesis of Natural Products," Vol. 2., Wiley, 1973
2. D.J.J. de Villiers, C.F. Garbers, and R.N. Laurie, *Phytochemistry*, 1971, 10, 1359
3. cf. R.C. Cookson and R.M. Tuddenham, *J. Chem. Soc. Chem. Comm.*, 1973, 742

4. T.G.H. Jones and F.B. Smith, J. Chem. Soc., 1925, 2530; 1926, 2767;
E.E. Boehm, V. Thaller and M.C. Whiting, J. Chem. Soc., 1963, 2535;
P. Teisseire and B. Corbieu, Recherches, 1969, 17, 5
5. A. St. Pfau and P. Plattner, Helv. Chim. Acta, 1934, 17, 129;
G.S. Krishna Rao, S. Dev, and P.C. Guha, J. Indian Chem. Soc., 1952, 29, 721
M.I. Golyaev, A.D. Dimbitskii, T.E. Serkebaeva and G.I. Krotova, Izv. Akad. Nauk Kaz. SSR Ser. Khim., 1969, 19, 48; Chem. Abs., 1969, 71, 64019;
R.J. Crawford, W.F. Erman, C.D. Broaddus, J. Amer. Chem. Soc., 1972, 94, 4298;
J. Alexander and G.S. Krishna Rao, Indian J. Chem., 1973, 11, 859
6. R.B. Bates, M.J. Onore, S.K. Pakinkar, C. Steelink, and E.P. Blanchard,
J.C.S. Chem. Comm., 1967, 1037;
J.J. Beereboom, J. Amer. Chem. Soc., 1963, 85, 3525; J. Org. Chem., 1965, 30, 4230
7. W.F. Erman, R.S. Treptow, P. Bakuzis and E. Wenkert, J. Amer. Chem. Soc., 1971, 93, 657

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